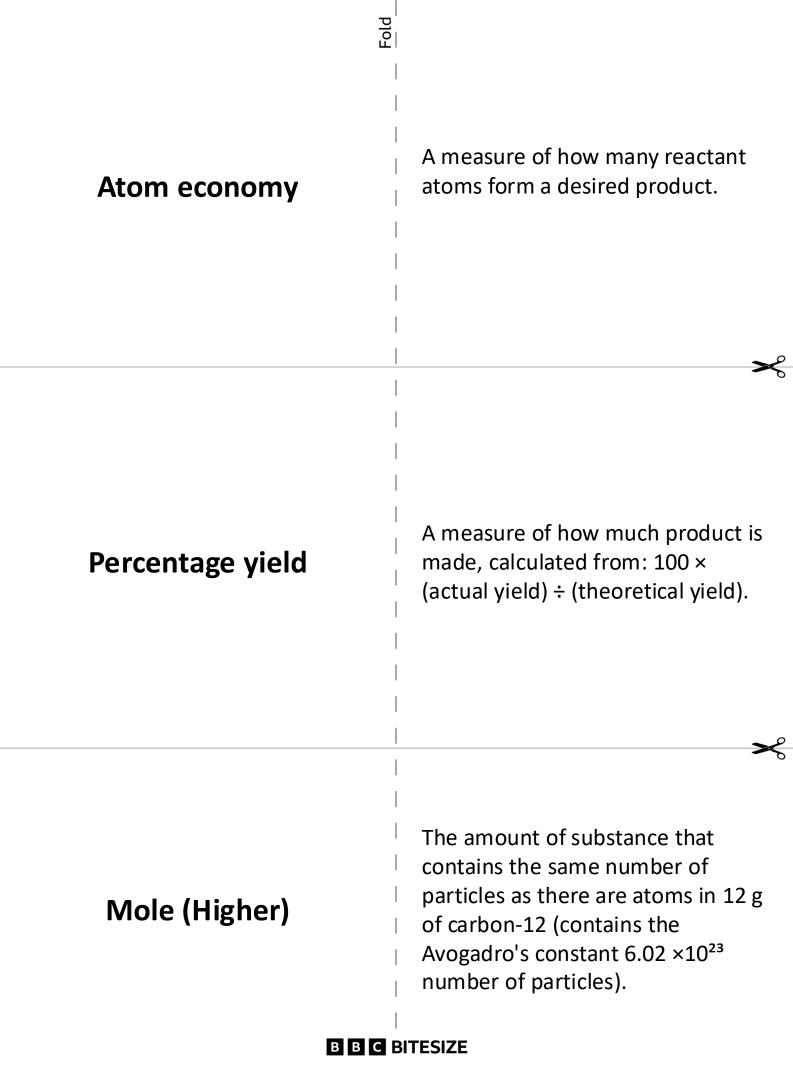


Ionic compound	An ionic compound occurs when a negative ion (an atom that has gained an electron) joins with a positive ion (an atom that has lost an electron).
Relative formula mass	The sum of the relative atomic masses of the atoms in a chemical formula.
Non-enclosed system	A system in which gases or other substances can enter or leave during a reaction.

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Uncertainty	The interval within which the true value of a quantity can be expected to lie.
Range	A measure of spread found by subtracting the smallest number from the biggest number. In other words, the difference between the highest and lowest values in a set of data.
Resolution	For a measuring instrument, the smallest change in a quantity that gives a change in the reading that can be seen.



Avogadro constant (Higher)	The number of particles in one mole of particles, 6.02 × 10 ²³ per mole.
Concentration (Higher)	A measure of the mass or amount of solute dissolved in a given volume of solvent or solution.
Limiting reactant (Higher)	The reacting substance that is completely used up in a chemical reaction, which determines how much product is made.